HOT WATER HEATING SYSTEMS By

F. E. Gieseke, College Architect

anytime-Day or Night to 12 p.m. Mrs. Parkhill's Across from Aggieland Pharmacy

Good Coffee

And Sandwiches



42 out of 54 colleges choose this FAVORITE pipe tobacco



and Yale agrees

TOOK UP at the windows of ___ Harkness to find out what the Yale man smokes. In the springtime you'll see him sitting in his window seat with a pipeful of Edgeworth between his teeth.

On Chapel Street . . . out at the Bowl ... everywhere the Yale man goes, his pipe and Edgeworth go with him. And at 42 out of 54 of the leading colleges and universities Edgeworth is the favorite tobacco.

A tobacco must be good to win the vote of so many discriminating smokers. And Edgeworth is good. To convince yourself try Edgeworth. You can get it wherever tobacco is sold ... 15¢ a tin. Or, for a generous free sample, write to Larus & Bro. Co., 105 S. 22d St., Richmond, Virginia.

EDGEWORTH SMOKING TOBACCO

Edgeworth is a blend of fine old burleys, withits natural savor enhanced by Edge-worth's distinctive eleventh process. Buy Edgeworth any-where in two forms -- "Ready-Rubbed" and "Plug Slice." All sizes, 15¢ pocket package to pound humidor tin.

EDGEWORTH EXTRA HIGH GRADE READY RUBBED EDGEWORTH EXTRA PLUG GRADE PLUG SLICE EDGEWORTH

C TALLER

The Estimated Cost Of Heating Hart | The next step in the calculation is Hall

building for a given period of time it to extend from October 1 to May 1, a s necessary to know the quantity of period of seven months. The average heat lost by the building during that mean temperature for this period at period and the cost of producing heat and of delivering it to the building. Heat College Station for the past 29 years was 59 degrees, according to the U.

and of delivering it to the building. In determining the heat lost by a given building, Hart Hall, for ex-ample, it is customary to assume that the indoor temperature should be maintained at 70 degrees, and that the lower, the mean indoor and outdoor heat lost by the building is propor-tional to the differences of the indoor degrees for the seven-month heating and outdoor temperatures. For ex-ample, for indoor and outdoor tem-peratures of 70 degrees and 25 de-which must be supplied to a building grees, and of 70 degrees and 55 de- during a period of 24 hours when the grees, the respective temperature outdoor temperature is one degree differences are 45 degrees and 15 de-below 65 degrees the total quantity of grees and it is assumed that the heat heat to be supplied to the building at loss of the building is three times as College Station will be 212×6 , or 1272large in the former as in the latter case

flow through various types of build- in general use for calculating the heat ing materials and it is therefore com-paratively easy to calculate the heat According to a diagram published in requirements for different localities. paratively easy to calculate the heat losses of buildings under given condi-tions. For example, Hart Hall will lose about 850,000 B. t. u. per hour when the outdoor temperature is 25 degrees, the hall temperature 55 de degrees, the hall temperature 55 de- Texas, 2,500, Oklahoma, 3,000, Kangrees and the room temperature 70 degrees

In addition to the heat lost by a building through the walls, floors and roof, heat is also lost with the air which is used for ventilation, and which flows through the building, entering at the outdoor temperature and leaving at the indoor temperature.

At 70 degrees one B. t. u. will heat about 55 cubic feet of air one degree. If we assume for Hart Hall that every occupant should be provided with 1400 cubic feet of outdoor air per hour, the heat loss for ventilation will be about 350,000 B. t. u. per hour when the outdoor temperature is 25 degrees; the total heat loss for the building will, therefore, be about 1,200,000 B. t. u. per hour.

The most efficient method of securing the ventilation referred to is to lower the upper sash and to raise the lower sash. When that is done outdoor air will flow o the room through the lower of the window When that is done and indoor air wii. flow out through the upper part of the window, and the outdoor atm.cspheric pressure will be equal to the indoor atmospheric pressure at an elevation at or near the center of the height of the window. The zone in which the two atmospheric pressures are equal is the neutral zone. The flow of the air through the windows can be demonstrated by a simple calculation. Let us assume that the windows are six feet high, that the neutral zone is at the center of the height of the windows, that 70 degrees air weighs 75 pounds and 25 degrees air 82 pounds per 1,000 cubic feet. If the atmospheric pressure at the neutral zone is W pounds per square foot, the outdoor pressure three feet above the neutral zone will be W—3 x .082 and the indoor pressure will be W—3 x .075. The indoor pressure is there-fore 0.007 pounds per square foot greater than the outdoor pressure; this excess pressure causes the air to flow outward through the upper part of the window. A similar calculation will show that at the bottom of the window the outdoor pressure is about 0.007 pounds per square foot greater than the indoor pressure and will cause the air to flow inward at the bottom of the window.

Knowing that the heat loss of Hart Hall under normal conditions 1,200,000 B. t. u. when the outdoor temperature is 25 degrees, the heat loss for any other outdoor temperature can be determined by proportion. For example, when the outdoor temperature is 55 degrees, the heat loss will be 400,000 B. t. u. per hour.

to determine the probable total heat requirement for the heating season. The period during which heating may be required varies with the latitude. To determine the cost of heating a For College Station we may assume it

THE BATTALION

degree-days.

The concept "degree-day" just de-It is known from experiment and scribed was first proposed by the from experience how much heat will American Gas Association. It is now sas, 5,000, Nebraska, 6,500, South Dakota 7,500, North Dakota, 9,500.

I believe that the heating load, 1272 degree-days calculated above for College Station, is too small for two reasons. First, because I believe w should reckon our degree-days with 70 degrees as the basis instead of with 65 degrees, and second, because the mean temperature cited above is the mean of the daily maxima and daily minima and this does not represent the mean temperature for which heat must be supplied. On many days from October 1 to May 1 the maximum temperature at College Station is above 70 degrees. I believe these high temperatures should be replaced by 70 degrees before the mean temperature for heating purposes is calculated. For example, if we consider the heating season October 1, 1928, to May 1, 1929, and campare the actual mean temperatures with those found by substituting 70 degrees for all maximum temperatures higher than 70 degrees we secure the following monthly and seasonal mean temperatures

Oct.	73.8	56.5	
Nov.	57.4	55.0	
Dec.	49.8	49.1.	
Jan.	50.7	49.8.	
Feb.	44.8	44.4.	
Mar	. 63.3	57.4.	
Apr.	. 71.0	60.0.	
Seas	son 59.	7 53.2.	
And w	e find th	at the mear	1 1

And we find that the mean tempera-ture for heating purposes is 53.2 di-



CURRY'S PLACE The Root Beer Stand

WE SERVE THE BEST EATS AND DRINKS

Next to Blue Moon Filling Station

period.

plied to the 29-year period referred stead of 2671. to above, the desired mean tempera- It was show cure is 0.888 x 59, or 52.4.

of 1272 as calculated above. However,

vided by 59.7, or 0.888 of the actual if the number of degree-days are calmean temperature for this particular culated with 70 degrees instead of 65 degrees as the basis, the number will If the same percentage may be ap- be 212 times 70 minus 52.4, or 3731 in-

It was shown above that the heat loss of Hart Hall is 1,200,000 B. t. u. The number of degree-days which should then be used for heat calcula-ture is 25 degrees. If the outdoor tions at College Station will be 212 temperature were to remain con-times 65 minus 52.4, or 2671, instead stant at 25 degrees for an entire day (Continued on page 3)

> Shirts and Breeches Blouses and Slacks Cleaning Pressing and Alterations a Specialty WIN OR LOSE - THE AGGIES FOR US. FRANK ZUBIK, Prop. The Ideal Gift For Christmas YOUR PHOTOGRAPH Kodak Finishing **Picture Frames Aggieland Studio**

AGGIELAND TAILOR SHOP

TAILOR MADE

JOE SOSOLIK, PROP.

1855 · SEVENTY-FIFTH ANNIVERSARY · 1930

91% of the energy we use demands VALVES

"Eighty-seven per cent of the energy we use in our daily life ... heat energy as well as mechan-ical energy, exclusive of that produced in our own bodies and brains... is derived from the hy-drocarbon chain, coal, oil, and gas. Water power yields 4%, firewood 6%, work animals 3%." GEORGE OTIS SMITH, U. S. Geological Survey

Take away the 87% of energy now extracted from coal, oil, and gas . . . and we would be back in the year 1855... the year Crane Co. was founded. Take valves and fittings away, and we would be deprived not of 87 but of 91%. For from water power as well as from coal, gas, and oil, energy is almost never extracted in the modern world but valves and fittings enter into the process.

It is significant that the history of Crane Co. and the history of modern utilization of natural energy, cover almost exactly the same period. Many years ago, Crane metallurgists and engineers began the development of piping materials for each new need as it appeared. The years since have seen every Crane resource ... research, engineering, production . . . devoted to supplying materials that would keep the road to progress open.

What Crane has learned and the materials that it has developed will be of vital interest to you after you leave school. Let us send you the story of research in piping metals, "Pioneering in Science."

